## NOVEL WATER-SOLUBLE TECHNETIUM(I) TRICARBONYLHYDROXYFLUORIDE

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On heating a mixture of HCOOH, 45 % HF, and KTcO<sub>4</sub> in an autoclave at 200°C and 120-150 atm of CO new [Tc<sub>4</sub>(CO)<sub>12</sub>F<sub>3</sub>OH] and known [Tc<sub>4</sub>(CO)<sub>4</sub>OH]<sub>4</sub> [1] were formed. The ratio of these complexes depends on the HCOOH:HF ratio. The fluoride complex occurs in the mother liquor and technetium tricarbonylhydroxide in the precipitate contaminated with  $TcO_2$  by-product. No other carbonylfluoride complexes were obtained. The mother liquor was evaporated under reduced pressure. [Tc<sub>4</sub>(CO)<sub>12</sub>F<sub>3</sub>OH] was isolated from the solid residue by vacuum sublimation.

[Tc<sub>4</sub>(CO)<sub>12</sub>F<sub>3</sub>OH] is yellow air-stable crystals soluble in water and polar organic solvents. This compound sublimes in vacuum at 210-230°C with partial decomposition. The IR spectrum of [Tc<sub>4</sub>(CO)<sub>12</sub>F<sub>3</sub>OH] in the carbonyl region contains two bands at 2048 cm<sup>-1</sup> and 1930 cm<sup>-1</sup> that is typical for Tc(CO)<sub>3</sub> facconfiguration. There are two sharp bands at 3650 and 3620 cm<sup>-1</sup> in the region of OH vibration. The composition of this compound was confirmed by elemental analysis for Tc and F.

As shown by powder X-ray diffraction, the crystal structure of  $[Tc_4(CO)_{12}F_3OH]$  considerably differs from that, of  $[Tc(CO)_3OH]_4$  [1]. The single crystal X-ray study revealed that  $[Tc_4(CO)_{12}F_3OH]$  crystallized in the orthorhombic cell, space group Pnma, a = 14.315(9), b = 14.030(10), c = 9.874(5) Â (at -125°C). The positions of F- and OH- ions in the cubane-like complex seem to be disordered. The main (average) interatomic distances are:  $Tc_7(O, F) = 2.20$  Â,  $Tc_7(C) = 2.20$  Å,  $Tc_7(C) = 2.20$  Å,

## References

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