



TECHNETIUM(I) HEXACARBONYL CATION AND ITS REACTIVITY

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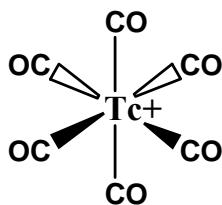
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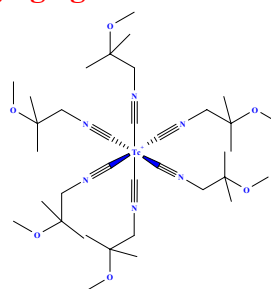
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**Techtium(I) hexacarbonyl cation $[\text{Tc}(\text{CO})_6]^+$ is a an analog
of well known heart imaging agent MIBI**



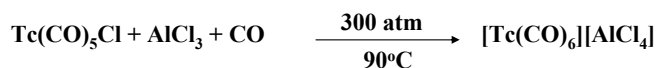
technetium(I) hexacarbonyl cation



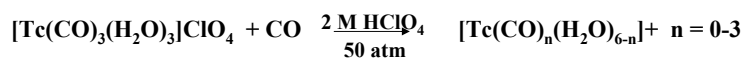
MIBI
hexakis(2-methoxy-2-
methylpropyl-1-isonitrile-
technetium(I)



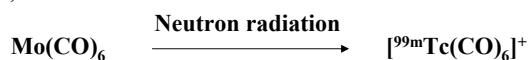
Published procedures for preparing technetium(I) hexacarbonyl



Hieber W., Lux F., Herget C. Ueber Kohlenoxidverbindungen des Technetiums // Z. Naturforsch., 1965, Vol. 20b, P. 1159-1165.



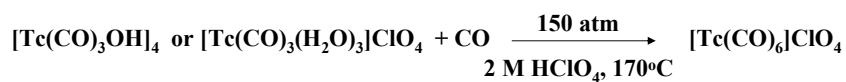
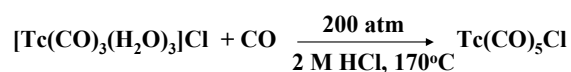
Aebischer N., Schibli R., Alberto R., Merbach A.E. Complete Carbonylation of *fac*-[Tc(H₂O)₃(CO)₃]⁺ under CO Pressure in Aqueous Media: A Single Sample Story! // Angew. Chem. Int. Ed., 2000, Vol. 39, No. 1, P. 254-256.



Nefedov, V.D. and Mikulaj, Stabilization of daughter ^{99m}Tc after β-decay of ⁹⁹Mo in molybdenum hexacarbonyl, Radiochemistry, 1973, Vol. 15, № 6, P. 846-852



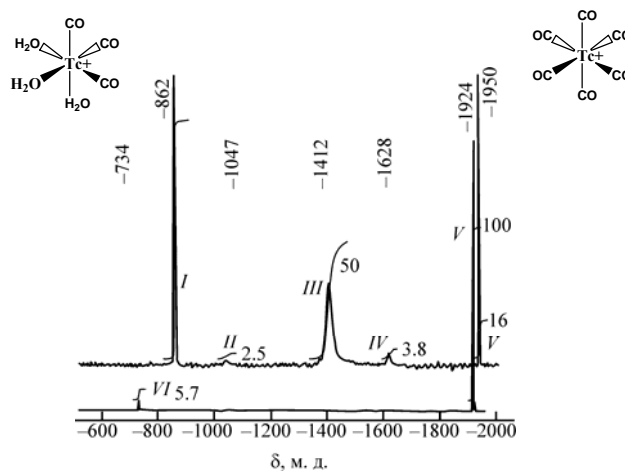
Synthesis of technetium hexacarbonyl



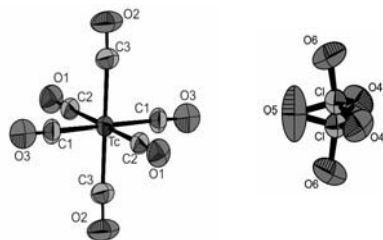
Autoclave for high-pressure carbonylation



⁹⁹Tc NMR spectrum of reaction mixture [Tc(CO)₃OH]₄ + CO в 2 M HClO₄



Molecular structure of [Tc(CO)₆]ClO₄



Interatomic distances

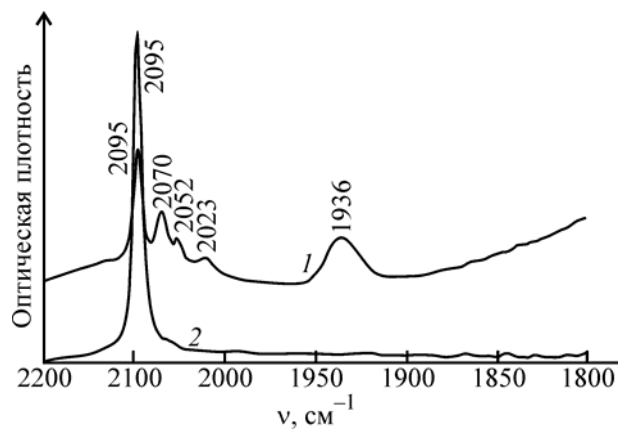
Tc-C1	2.032(3)	C1-O3	1.109(4)
Tc-C2	2.029(3)	C2-O1	1.111(4)
Tc-C3	2.024(3)	C3-O2	1.117(4)
Cl-Cl	0.728(3)	Cl-O5	1.358(6)
Cl-O4	1.393(4)	Cl-O6	1.419(6)

Bond angles

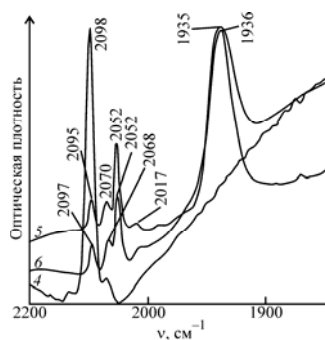
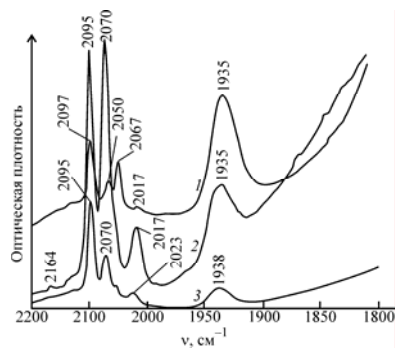
C1-Tc-C1	180	C2-Tc-C3	91.47(13)
C1-Tc-C2	90.35(12)	C2-Tc-C3	88.53(13)
C1-Tc-C2	89.65(12)	C3-Tc-C3	180
C1-Tc-C3	91.41(14)	O3-Cl-Tc	178.2(3)
C1-Tc-C3	88.59(14)	O1-C2-Tc	179.6(3)
C2-Tc-C2	180	O2-C3-Tc	177.3(3)



IR Spectra of (1) reaction mixture $[\text{Tc}(\text{CO})_3\text{OH}]_4 + \text{CO}$ in 2 M HClO_4 and (2) solution of white crystals of $[\text{Tc}(\text{CO})_6]\text{ClO}_4$ in acetonitrile

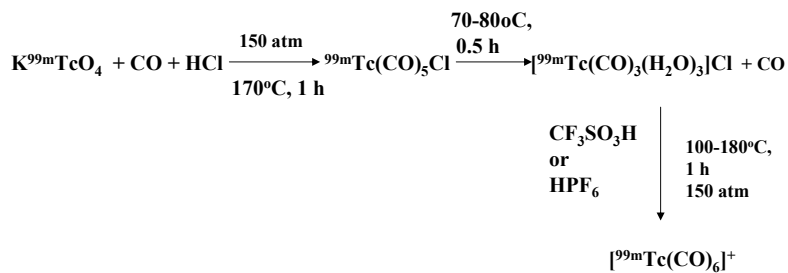


IR spectra of reaction mixtures of $[\text{Tc}(\text{CO})_6]^+$ synthesis in 1 – CF_3COOH (2 моль/л, 150 atm CO, 100°C, 10 h), 2 – H_2SO_4 (2 M, 100 atm CO, 200°C, 4 h), 3 – HOSO_2CF_3 (2 M, 150 atm CO, 170°C, 1 h), 4 – HPF_6 (2 M, 140 atm CO, 175°C, 1 h), 5 – HF (2 M, 150 atm CO, 175°C, 2 h), 6 – HPF_6 (0.5 M, 110 atm CO, 175°C, 1 h).





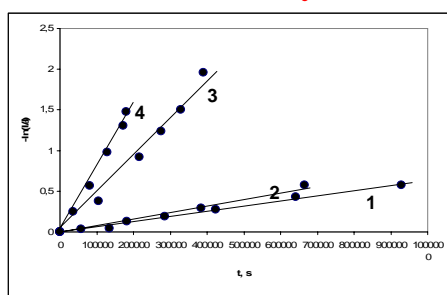
Preparation of $[\text{}^{99\text{m}}\text{Tc}(\text{CO})_6]^+$



Yield of $[\text{}^{99\text{m}}\text{Tc}(\text{CO})_6]^+$ in the presence of triflic acid or HPF_6 is 15 and 40%, respectively.



Kinetics of technetium hexacarbonyl decarbonylation

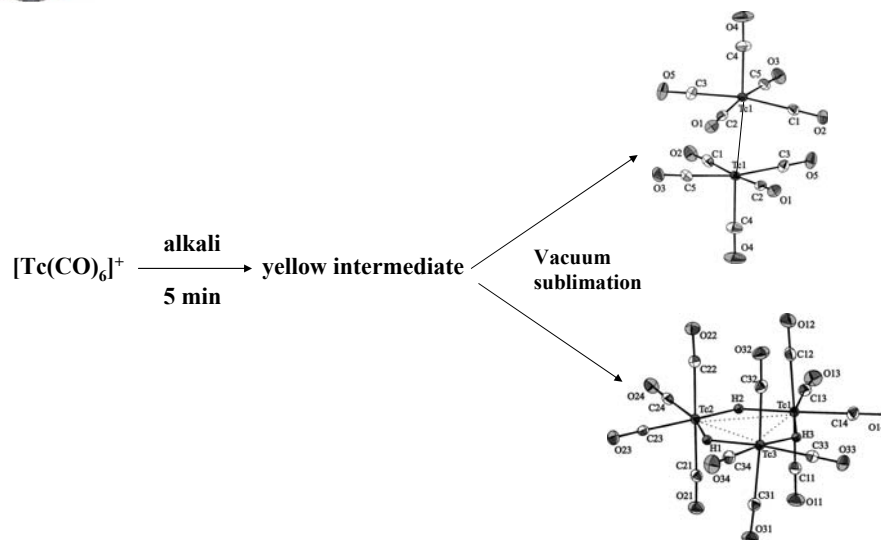


Semilogarithmic anamorphosis of kinetics curves of hexacarbonyl decarbonylation in acetonitrile at (1) 31.4, (2) 37.1, (3) 44.8 and (4) 48.5°C.

Rate constant of decarbonylation in CH_3CN : $[\text{Tc}(\text{CO})_6]^+$ ($6.53 \times 10^{-7} \text{ s}^{-1}$, 304.4 K) and $\text{TcBr}(\text{CO})_5$ ($2.65 \times 10^{-5} \text{ s}^{-1}$, 303.4 K).



Hydrolysis of technetium hexacarbonyl



Conclusions

- Technetium hexacarbonyl perchlorate was prepared and characterized by IR, NMR spectroscopy and SCXRD.
- A procedure for preparing $[\text{}^{99\text{m}}\text{Tc}(\text{CO})_6]^+$ was developed and this carrier free complex was obtained in reasonable yield.
- Technetium(I) hexacarbonyl cation is kinetically stable with respect to thermal decarbonylation in solutions and rapidly decomposes in the presence of strong bases.